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An Introduction to Blood Gas Analysis

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Things You Can Know Without Performing a Blood Gas Analysis

- Historical
 - Is the patient having difficulty breathing?
 - Is the patient having a change in symptoms over time?
- Physical
 - Is the patient working to breathe?
 - Is the patient wheezing?
 - Does the patient have rales?
- Noninvasive Measurement
 - Is the patient sufficiently oxygenated (SaO_2)?

Pulse Oximetry ‘The 5th Vital Sign’

- Non-invasive
- Instantaneous
- Ubiquitous
- $\text{SaO}_2 < 95\%$ the usual cutoff for normal versus ‘abnormal’
- Limitations:
 - Patient must have pulse
 - Detects only significant decreases in PO_2
 - Does not comment on content



 Roger W. Stevens ([Wikipedia](#))

Reasons to Sample Arterial Blood

- Firmly establish the severity of an oxygenation abnormality
- To evaluate hyper- or hypoventilation
 - Currently no convenient noninvasive way of evaluating $p\text{CO}_2$
- To determine acid-base status, particularly in patients with metabolic acidosis (e.g., diabetic ketoacidosis)
- To track the application of mechanical ventilation in a critically ill patient

Most Dyspneic Patients Don't Require ABG Analysis

- When cause of dyspnea is established
 - Asthma, CHF, restrictive lung disease, etc.
- When cause of dyspnea is suspected and patient is not especially ill
 - E.g., a child with new-onset wheezing in January
- When dyspnea is so severe as to warrant immediate mechanical ventilation
 - The decision to intubate and mechanically ventilate is almost always one based on clinical, not laboratory, grounds

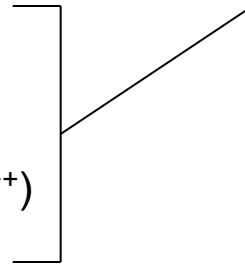
Limitations of ABGs

- ABGs measure gas partial pressures (tensions)
 - Remember: PO_2 is not the same as content! A severely anemic patient may have an oxygen content reduced by half while maintaining perfectly acceptable gas exchange and therefore maintaining pO_2
- Technical issues
 - They hurt
 - Sampling from a vein by mistake
 - Finding an arterial pulse can be difficult in very hypotensive patients
 - Complications such as arterial thrombosis are possible, but awfully rare



ABGs: What You Get

- Arterial PO_2
- Arterial PCO_2
- Arterial pH
- Some electrolytes (e.g., Na^+ , K^+ , Ca^{++})
- Lactate



Measured.
The real
meat of the
sample

- $[HCO_3^-]$
- SaO_2
- Other assorted calculated results

An Organized Approach to ABG Interpretation

- Determining oxygenation abnormalities
- Determining acid-base status and evaluating adequacy of ventilation

Evaluating Oxygenation

- What is a 'normal' PO₂?
 - Oxygenation gradually deteriorates during life
 - Several calculations available for determining 'normal' based on patient age.

$$PaO_2 = 104.2 - (0.27 \times \text{age})$$

i.e., 30 year old ~ 95 mmHg
60 year old ~ 88 mmHg

- Note: Some patients with previous (and now resolved) severe pulmonary diseases may never recover their full lung function, so any sense of 'normal' needs to be tempered with historical information

Oxygenation: Two Key Questions Addressed with an ABG

Is the patient hypoxic?

Is the hypoxia due to:

Hypoventilation

One of the 3 other causes

Or a combination of both

Importantly, an ABG alone cannot differentiate diffusion block, V/Q inequality, and shunt!

Looking at the PO_2 versus Calculating the A-a gradient

- In a comfortable patient breathing room air, glancing at the PO_2 will allow a cursory interpretation of oxygenation
- However, most ABGs are performed in sick patients
 - Supplemental oxygen may be present
 - Importantly, the patient may be compensating for an oxygenation defect by hyperventilating, hiding the abnormal oxygenation

Evaluating Oxygenation with ABGs

- Step 1. Determine the A-a gradient

A-a Gradient =

$$[(P_{\text{atm}} - P_{\text{H}_2\text{O}}) \times \text{FiO}_2] - (P_{\text{CO}_2}/\text{RQ}) - P_{\text{aO}_2}$$

$$P_{\text{atm}} = 760 \text{ mmHg}$$

$$P_{\text{H}_2\text{O}} = 47 \text{ mmHg}$$

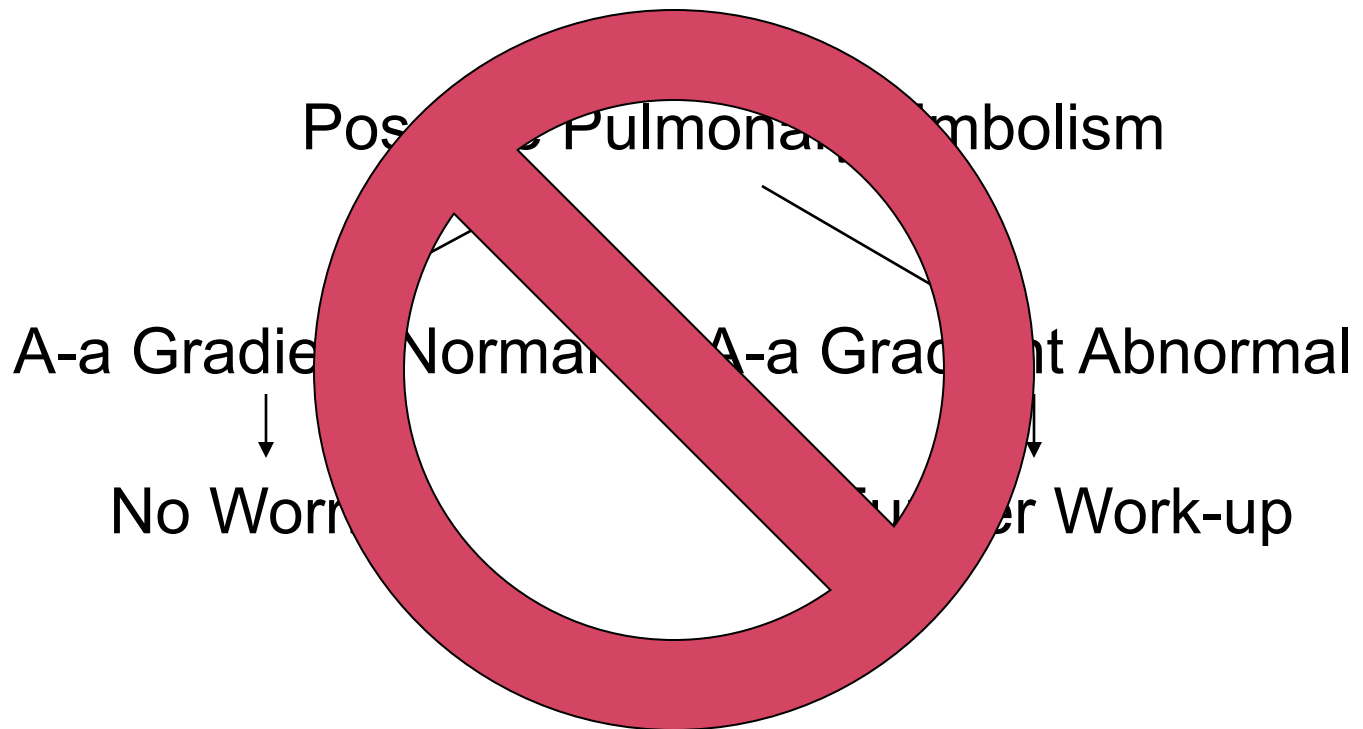
$\text{FiO}_2 = 0.21$ on room air at sea level

P_{CO_2} is taken from the blood gas measurement

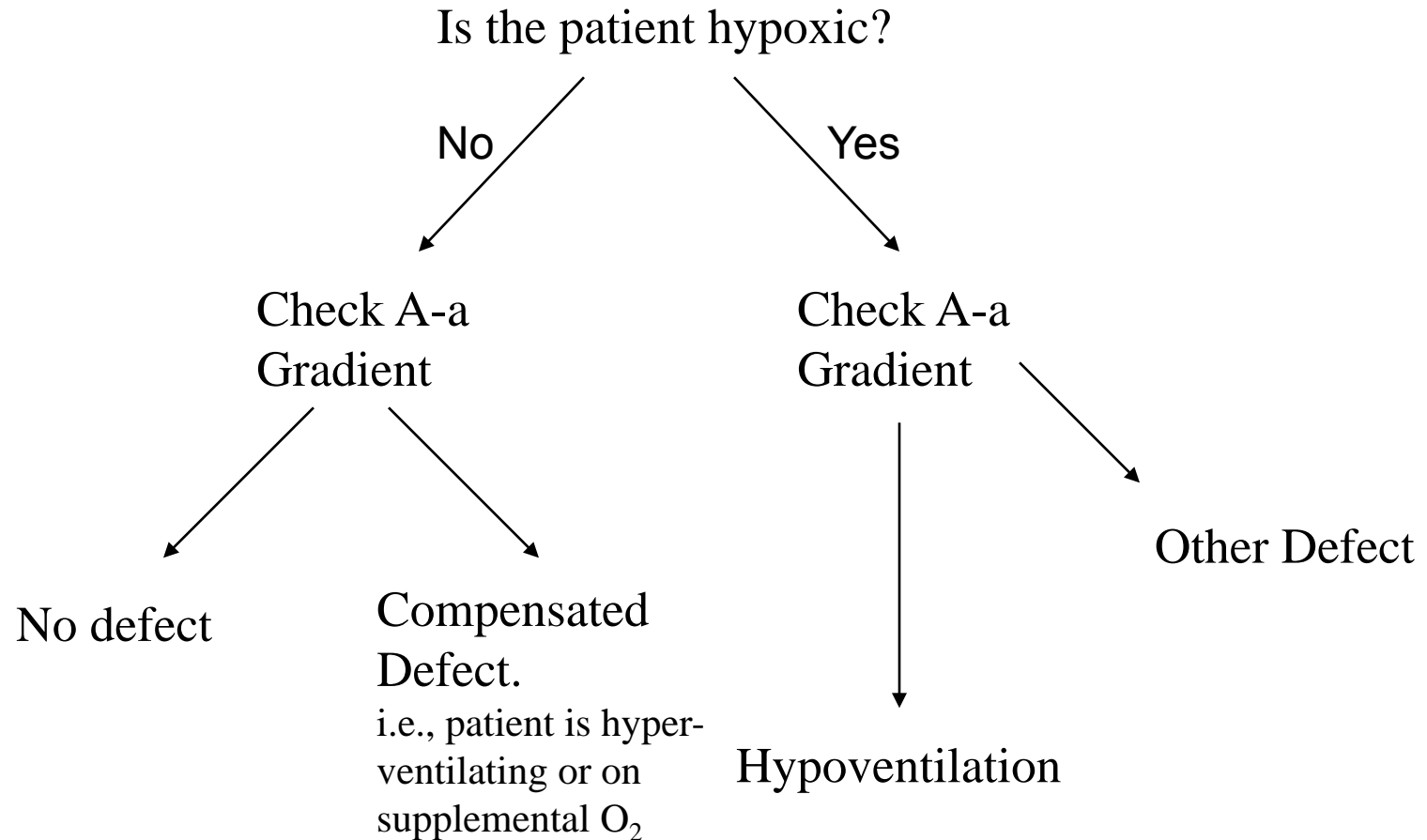
RQ can be assumed to be 1 (possible range from 0.7 to 1.0)

Why do details like RQ matter?

- ABGs occasionally used as dichotomous results, prompting changes in management



Evaluating Oxygenation with ABGs



Evaluating Acid-Base Status

- Both metabolic and respiratory abnormalities can alter pH
- Respiratory and renal function strive to keep pH = 7.4.
 - Minute ventilation can respond very quickly to metabolic acid-base problems
 - Renal excretion or retention of bicarbonate takes days to compensate for respiratory acid-base problems
- For our purposes, 3 questions:
 - Is the abnormality respiratory or metabolic?
 - If respiratory, is it acute or chronic?
 - If metabolic, is the respiratory system responding appropriately?

Evaluation of Acid-Base Status: Is the patient acidemic or alkalemic?

What is the pH?

< 7.38



Acidemic

> 7.42



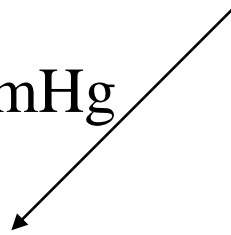
Alkalemic

Evaluation of Acid-Base Status: Is the disorder respiratory or metabolic?

If acidemic ($\text{pH} < 7.38$)

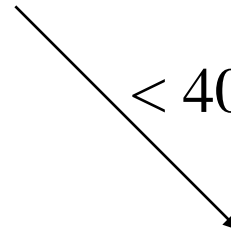
What is the PCO_2 ?

$> 40 \text{ mmHg}$



Respiratory acidosis

$< 40 \text{ mmHg}$



Metabolic acidosis

Evaluation of Acid-Base Status: Is the disorder respiratory or metabolic?

If alkalemic ($\text{pH} > 7.42$)

What is the PCO_2 ?

$> 40 \text{ mmHg}$

```
graph TD; A["If alkalemic (pH > 7.42)"] --> B["What is the PCO2?"]; B -- "> 40 mmHg" --> C["Metabolic alkalosis"]; B -- "< 40 mmHg" --> D["Respiratory alkalosis"];
```

$< 40 \text{ mmHg}$

Metabolic alkalosis

Respiratory alkalosis

For respiratory abnormalities, is the condition acute or chronic?

Acute respiratory disturbances change **pH 0.08** units for every **10 mmHg** deviation from normal

Therefore, in acute respiratory acidosis, the pH will fall by $0.08 \times [(PCO_2 - 40)/10]$

In acute respiratory alkalosis, the pH will rise by $0.08 \times [(40 - PCO_2)/10]$

For respiratory abnormalities, is the condition acute or chronic?

Chronic respiratory disturbances only change **pH 0.03** units for every **10 mmHg** deviation from normal

Therefore, in chronic respiratory acidosis,
pH will fall by $0.03 \times [(\text{PCO}_2 - 40)/10]$

In chronic respiratory alkalosis, the
pH will rise by $0.03 \times [(40 - \text{PCO}_2)/10]$

Regarding Metabolic Acidosis

- It is common for patients with severe respiratory disease to at some point develop other systemic illnesses producing metabolic acidosis.
- Patients with metabolic acidosis will attempt to hyperventilate to correct their pH
- It's useful in patients with lung disease to determine how successful they are in 'blowing off their CO₂'

Appropriateness of Respiratory Response to Metabolic Acidosis

$$\text{Predicted Change in PCO}_2 = (1.5 \times \text{HCO}_3) + 8$$

If patient's PCO₂ is roughly this value, his or her response is appropriate

If patient's PCO₂ is higher than this value, they are failing to compensate adequately

Example 1

A 59 year old with a week of upper respiratory symptoms followed by one day of fever, chest pain, and dyspnea on exertion

$$\text{pH} = 7.48$$

$$\text{PCO}_2 = 28 \text{ mm Hg}$$

$$\text{pO}_2 = 54 \text{ mmHg}$$

pH 7.48, PCO₂ 28, PO₂ 54

What is his A-a gradient?

$$[(760-47) \times 0.21] - (28/0.08) - 54 = 61 \text{ mmHg}$$

pH 7.48, PCO_2 28, PO_2 54

Is this a respiratory or metabolic alkalosis?

pH 7.48, PCO₂ 28, PO₂ 54

Is this an acute or chronic abnormality?

If acute, then pH change should be $0.08 \times [(40 - \text{PCO}_2)/10]$

$$0.08 \times [(40-28)/10] = 0.09, \text{ or a pH of } 7.49$$

If chronic, then pH change should be $0.03 \times [(40 - \text{PCO}_2)/10]$

$$0.03 \times [(40-28)/10] = 0.03, \text{ or a pH of } 7.43$$

How would you interpret this ABG?

pH 7.48

PCO₂ 28

PO₂ 54

- Hypoxic
- Acute respiratory alkalosis

Example 2

A 47 year old woman with a 65 pack/year history of tobacco use is being evaluated for disability due to dyspnea

pH 7.36

PCO₂ 54 mmHg

PO₂ 62 mmHg

pH 7.36, PCO₂ 54, PO₂ 62

What is her A-a gradient?

$$[(760-47) \times 0.21] - (54/0.8) - 62 = 20 \text{ mmHg}$$

pH 7.36, PCO₂ 54, PO₂ 62

Is this a respiratory or metabolic acidosis?

pH 7.36, PCO₂ 54, PO₂ 62

Is this an acute or chronic abnormality?

If acute, then pH change should be $0.08 \times [(\text{PCO}_2 - 40)/10]$

$$0.08 \times (54-40)/10 = 0.11, \quad \text{or a pH of 7.29}$$

If chronic, then pH change will be $0.03 \times [(\text{PCO}_2 - 40)/10]$

$$0.03 \times (54-40)/10 = 0.04, \quad \text{or a pH of 7.36}$$

How would you interpret this ABG?

pH 7.36

PCO₂ 54

PO₂ 62

- Hypoxic, with both a hypoventilatory and primary oxygenation abnormality
- Chronic respiratory acidosis

Want more?

For more cases,

<http://www.sitemaker.umich.edu/younger>

For the purposes of looking sharp on the wards (and for the exam)

- Be able to do these problems
 - Practice, practice, practice
 - Take advantage of PDA-based software
- Anticipate special circumstances
 - What if the patient isn't breathing room air?
 - More than one defect at a time (i.e., both respiratory and metabolic disease simultaneously. We will graciously save that one for the renal folks)

Additional Source Information

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Slide 9: Pemed, www.pemed.com/lab/labanalz/labanalz.htm